

Unit 3: Electrostatics

Particle		Mass (kg)	Charge (C)
n ⁰	neutron	1.67 x 10 ⁻²⁷	0
p ⁺	proton	1.67 x 10 ⁻²⁷	+1.6 x 10 ⁻¹⁹
e ⁻	electron	9.11 x 10 ⁻³¹	-1.6 x 10 ⁻¹⁹

$k = 9 \times 10^9 \text{ N} \cdot \text{m}^2/\text{C}^2$ $1 \mu\text{C} = 1 \times 10^{-6} \text{ C}$ *atomic # = # protons in nucleus*

$$F_g = \frac{G \cdot m_1 \cdot m_2}{r^2} \qquad F_e = \frac{k|q_1 \cdot q_0|}{r^2}$$

$$\vec{E} = \frac{k \cdot q_1}{r^2} \qquad F_e = q_0 \cdot \vec{E} \qquad V = \frac{k \cdot q_1}{r}$$

$$PE_e = KE = \frac{1}{2} m \cdot v^2 \qquad W = PE_e = q_0 \cdot \Delta V = \frac{k|q_1 \cdot q_0|}{r}$$